Growth and Characterization of Manganese (II) Sulphate and L-Lysine doped Manganese (II) Sulphate (LMnSO₄) Crystals

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Abstract: Single crystals of Manganese (II)sulphate doped with basic amino acid L-Lysine were grown successfully by slow evaporation method at ambient temperature. The concentration of dopants in the mother solution was varied from 1gm to 2gm. The solubility data for all dopant concentration were determined. The semiorganic material LMnSO₄ was synthesized and its structure was confirmed by powder X-ray diffraction study. Fourier transform infrared spectroscopy study confirms the incorporation of L-Lysine into MnSO₄ crystal. The doped crystals are optically better and more transparent than the pure ones having wide transmission spectra lying between 190 and 1100 nm. The dopant increases the hardness value of the material and it also depends on the concentration of the dopant. The dielectric loss of the material was calculated using dieletric measurement.

Keywords: Crystal morphology; X-ray diffraction; Growth from solutions; Manganese compounds;

1. INTRODUCTION

People grow crystals for two main reasons, to understand how crystals grow (aesthetic) and for the utility (scientific or technological applications of the grown crystals); for either of these, one must evaluate the quality of the crystals grown. The responsibility for the exquisiteness of the crystal is due to their structural simplicity, symmetry and purity. In recent years several studies deal with organic, inorganic and semiorganic molecules and materials due to the increasing need for cheap and easily processable materials for photonics applications. Semi organic crystals have the combined properties of both inorganic and organic crystals like high damage threshold, wide transparency range, less deliquescence and high non-linear co-efficients which makes them suitable for device fabrication ¹. In order to improve chemical stability, efforts have been made on amino acid with organic and inorganic compounds due to laser damage theroshold, and nonlinear and linear optical properties. Due to its properties and various applications in Bio-medical areas,L-Lysine was selected for the present work. Among the many functions of the amino acids, L-Lysine is capable to form collagen and repair tissues in the body. Like many metal sulphates, manganese (II) sulphate forms a variety of hydrates; monohydrate, pentahydrate and heptahydrate. The monohydrate is most common. In this present work, Manganese(II)sulphate and L-Lysine doped Manganese(II)sulphate crystals were synthesized and characterized by X-ray diffraction (XRD), Fourier Transform Infrared(FTIR), UV-Visible Dielectric study, and Microhardness techniques.

2. MATERIALS AND METHODS

The 100 ml of distilled water was measured and

taken in a cleaned beaker. 10g of Manganese (II) Sulphate (MnSO₄) Merck company-AR Grade was weighed and added with distilled water. The solution was allowed to stir using Magnetic stirrer. After 5 minutes MnSO₄ salt was dissolved in water. Temperature was maintained by hot plate attached with Magnetic stirrer and measured by thermometer. MnSO₄ was added repeatedly to the solution and stirred. The MnSO₄ solution reached saturated stage with 60 g of MnSO₄ salt added to water and the solution was stirred for two hours. The beaker contains saturated solution of MnSO₄ with PH-6 prepared at room temperature. This solution was allowed for slow evaporation at room temperature.Good quality single crystals of MnSO₄ were harvested after 23 days in different sizes, maximum up to 1.8cm x .9cm x .5cm which is shown in figure 1.a. For the growth of single crystals of L-Lysine doped manganese (II) sulphate, 1gram of L-Lysine was added to the saturated solution of $MnSO_4$ which was maintained at $50^{0}C$ to avoid decomposition. Then the solution was allowed to stir using magnetic stirrer for more than two hours. The solution was allowed to evaporate slowly at room temperature. LMnSO₄ salt was synthesized according to the reaction,

$$\begin{array}{c} MnSO_4 + HO_2CCH(NH_2)(CH_2)_4NH_2 \\ \downarrow \\ COO^-H^+ - CH - NH_2 - CH_2 - CH_2 - CH_2 - CH_2 - NH^+ - \\ Mn^{2+}SO_4^{2-} \\ + H_2SO_4 \end{array}$$

After 18 days, good quality single crystals were obtained which is named as LMnSO₄[A] in different sizes, maximum up to 1.15cm x0.3cm x0.1cm as shown in Figure 1.b. In the same way, rather than 1 gram of L-Lysine, 2 grams of L-Lysine was added to the saturated solution of MnSO₄. After 32 days the good quality crystals of LMnSO₄-[B] were grown in different sizes maximum size up to 0.7 cmx1cm x1cm.which is shown in Figure 1.c.



Figure 1.a. MnSO₄ single crystal.



Figure 1.b. MnSO₄[A] single crystal.



Figure1.c . LMnSO₄[B] single crystal.

3. CHARACTERIZATION

The grown crystals have been analyzed by different characterization techniques. The structure of grown single crystals of MnSO₄ and amino acid doped MnSO₄ was confirmed by powder crystal X-ray diffraction analysis using (SEIFERT XRD 3000P) nickel filtered CuK α radiation (36KV, 20mÅ, λ =1.5418). The functional groups were identified by using PERKIN ELMER RX1 Fourier Transform Infrared spectrophotometer in the range of 400-4000 cm⁻¹. The optical properties of the crystals were examined between 200 and 1100 nm using LAMBDA-35 UV-Vis spectrometer. The mechanical property of LMnSO4 crystal was studied by Vickers hardness test. The applied loads were 25, 50 and 100 grams.

3.1 Powder crystal X-ray diffraction analysis

Powdered samples of manganese (II) sulphate and L-Lysine doped manganese (II) sulphate semi organic crystals were subjected to powder X ray studies. In this, the strong observable peaks indicate the highly crystalline nature of the sample.Powder XRD reveals that intensity of the LMnSO₄[A]

and LMnSO₄[B] were found to be maximum in the direction [011]and[120] respectively . XRD patterns of the grown crystals are shown in figures 2a, 2b and 2c. The X-ray is reflected in the reflecting planes with the angle 15-70°. The powder X-ray diffraction studies have been carried out to confirm the crystallinity and to determine the lattice parameters of the grown sample. From the XRD data, it is observed that both Manganese(II)sulphate and L-Lysine doped Manganese(II)sulphate are orthorhombic. The calculated lattice parameter values of Manganese(II)sulphate and L-Lysine doped Manganese(II)sulphate are presented in table 1. The results of the present work are in good agreement with the reported values⁷. In the case of doped sample, a slight variation in the cell volume is observed. The axial angles were $\alpha = \beta = \gamma =$ 90°. This value is same for all grown crystals.



Figure 2.a. PXRD pattern of MnSO₄ single crystals.



Figure 2.b PXRD pattern of LMnSO₄[A] single crystals.



Figure 2.c. PXRD pattern of LMnSO₄[B] single crystals.

	r	r	
compound	MnSO ₄	LMnSO ₄ [A]	LMnSO ₄ [B]
a	14.86	0.3754	0.5297
b	7.303	1.140	2.104
с	6.67	0.33	0.5169
Cell volume	723.8	0.1412	0.5761
Average critical size nm	71	20	25
Average dislocation density 10 ¹⁵	0.1934	0.25	0.1502
Average strain	0.1878	0.1489	0.04871

Table 1: Lattice parameters of LMnSO₄ single crystals for various combinations

3.2 UV- Spectral Analysis

The UV - visible spectrum was recorded for the powdered sample of the crystals grown by slow evaporation . This study was carried out in the same spectral range for the grown $LMnSO_4$ crystals A and B .The recorded optical transmittance spectrum of the grown single crystals of $MnSO_4$, $LMnSO_4$ [A] and $LMnSO_4$ [B] are shown figures 3a, 3b and 3c.



Figure 3.a Optical transmittance spectrum of MnSO₄.



Figure 3.b Optical transmittance spectrum of MnSO₄ [A].



Figure 3.c Optical transmittance spectrum of LMnSO₄[B].

From above studies, the grown crystals have good optical transparency between 192 to 1100 nm. The lower cutoff value of grown crystals is above 300 nm.So the grown crystals have good optical transparency. The transmittance range of grown crystals is increasing with increase in L-Lysine percentage and the lower cut off values of grown crystals is also increasing with increase in L-Lysine percentage. So the grown crystal L-Lysine doped Manganese(II) sulphate is used for UV-applications.

3.3 FTIR analysis of grown cryatals

The FTIR spectrum was recorded for the powdered samples of the crystals grown by slow evaporation using attenuate total reflectance method in the frequency range 400-4000cm⁻¹ by a BRUKER 66 VFT-IR spectrometer. In order to make comparison,the same study was performed for $MnSO_4$ and $LMnSO_4$ crystals. The recorded FTIR spectrum for the grown crystals are shown in the figures 4.a, 4.b and 4.c. From the figures below, various absorption peaks present in the recorded FTIR spectrum for all grown crystals were assigned to their corresponding functional groups and are listed in table 2.



Figure 4.a FTIR analysis for MnSO₄ single crystals.

Table 2. FTIR spectral data of grown crystals with

standard values.

Mode of vibrations	Standard wave Number cm ⁻¹	MnSO ₄	LMnSO ₄ [A]	LMnSO ₄ [B]
O-H stretching	3400- 2400	3335	3399	3400
C-H bending	700- 610(b)	631	614	613
C-O Stretching	1260- 1000(S)	1101	1101	1106
O-H bending	1440- 1400	-	1490	-
Solfonate s=O stretching	1350- 11750	2180	2142	2141
S-O stretching	1000-750	777	-	-
Metal oxides bonding	600-500	464	-	-
Mode of vibrations	Standard wave number cm ⁻¹	MnSO ₄	LMnSO ₄ [A]	LMnSO ₄ [B]
O-H stretching	3400- 2400	3335	3399	3400
C-H	1			
bending	700- 610(b)	631	614	613
bending C-O Stretching	700- 610(b) 1260- 1000(S)	631 1101	614	613
bending C-O Stretching O-H bending	700- 610(b) 1260- 1000(S) 1440- 1400	631 1101 -	614 1101 1490	613 1106 -
bending C-O Stretching O-H bending Solfonate s=O stretching	700- 610(b) 1260- 1000(S) 1440- 1400 1350- 11750	631 1101 - 2180	614 1101 1490 2142	613 1106 - 2141
bending C-O Stretching O-H bending Solfonate s=O stretching S-O stretching	700- 610(b) 1260- 1000(S) 1440- 1400 1350- 11750 1000-750	631 1101 - 2180 777	614 1101 1490 2142 -	613 1106 - 2141 -



Figure 4.b FTIR for LMnSO₄[A].



Figure 4.c. FTIR FOR LMnSO₄ [B]

3.4 Vicker's Micro hardness study

Hardness is a measure of material's resistance to localized plastic deformation. It plays a key role in device fabrication. The mechanical property of LMnSO₄ crystal was studied by Vickers hardness test. The applied loads were 25, 50 and 100 grams. The measurement was done at different points on the crystal surface and the average value was taken as H_v for a given load.

The Vicker's micro hardness was calculated using the relation

$Hv = 1.8544 P / d^2$

Where, P - is the applied load and d- is the diagonal length of the indentation impression. The calculated Vickers hardness values for LMnSO₄ crystals as a function of load is shown in figures 5.a, 5.b & 5.c. Vickers Hardness value of LMnSO₄ crystal is grater than 1.6. It is concluded that the samples are soft materials.

VARIATIONS OF VICKER'S MICRO HARDNESS VALUES WITH APPLIED LOAD





Figure 5.a Hardness curve for MnSO₄ single crystal.







Figure 5.b Hardness curve for LMnSO₄[A] single crystal.



Figure 5.c Hardness curve for LMnSO₄[B] single crystal.



3.5 DIELECTRIC STUDY:

Optically good quality single crystals of $LMnSO_4$ were selected for dielectric measurements using LCR HITESTER. The selected samples were cut using a diamond saw and polished using paraffin oil. Silver paint was applied on both the faces to make a capacitor with the crystal as a dielectric material. The dielectric constant is calculated using the relation

 $D = Cd / \epsilon_0 A$

Where C is the capacitance, d is the thickness, A is the area and ϵ_0 is the absolute permittivity of free space (8.854 × 10⁻¹² F/m).

The variation of dielectric constant (D)

was studied as a function of frequency for the grown crystal and is shown in Figure 6.a The high value of dielectric constant at low frequencies may be due to the presence of all the four polarizations and its low value at higher frequencies may be due to the loss of significance of these polarizations gradually. From the figure 6.a, it is also observed that dielectric constant decreases with increase in frequency. The variation of dielectric loss with frequency is shown in Figure 6.b. The characteristics of low dielectric loss at very high frequency suggest that it possesses enhanced optical quality with lesser defects and this parameter is essential for nonlinear optical applications



Figure 6.a. Dielectric constant Vs log f.



Figure 6.b.s Dielectric loss (tan b) Vs log f.

4 CONCLUSIONS

The inorganic material $LMnSO_4$ was synthesized and its structure was confirmed by powder X-ray diffraction study. The size of the crystal depends on combinations of MnSO4 and L-Lysine. The [021],[011] and [120], facets are the most prominent among the other facets of the grown crystal. The determined lattice parameter values reveal that the grown crystals belong to orthorhombic system⁷. The functional groups present in the grown crystals were confirmed by FTIR spectroscopy in comparison with that of standard wavelength in the range 0f 190-1100nm.UV-vis study showed that the

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grown crystals have good optical transparency between 300-1100nm. The Vickers hardness number of the grown crystal increase with load at lower load conditions and then saturates. The calculated value of Meyer's index 'n' of the crystals is greater than 1.6 and reveals that they are soft. The dielectric measurements reveal that LMnSO₄ crystal possesses enhanced optical quality with lesser defects.

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